

BSN NURSING ENTRY TEST 2024 PAPER
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6. According to the author, a serious art critic may avoid being prejudiced by her subjective reactions if she:

- A. treats an artwork in relation to a fixed set of ideas and values
- B. brings to her observation a knowledge of art history and aesthetic theory
- C. allows more time for the observation of each artwork
- D. takes into account the preferences of other art critics

7. The author implies that it is insufficient to treat a work of art solely historically because:

- A. doing so would lead the critic into a dilemma
- B. doing so can blind the critic to some of the artwork's unique qualities
- C. subjective reactions can produce a biased response
- D. critics are not sufficiently familiar with art history

Four lettered pairs (A to D) follow a related pair of words given in capitals. Select the lettered pair that best expresses a relationship similar to that expressed in the original pair in capitals.

8. CONDUIT: WATER::

9. CARBOHYDRATES: GREASY::

- A. aversive: regression
- B. sugar: cavities
- C. pressure: extrusion
- D. hostility: war

10. ALWAYS: NEVER::

- A. often: rarely
- B. frequently: occasionally
- C. constantly: continually
- D. intermittently: casually

BIOLOGY

11. Human kidney contains _____ pyramids.

- A. 12-16
- B. 10-13
- C. 17-20
- D. 5-10

12. The presence of stones in the human kidney is called _____

- A. Cement
- B. Nephrolithiasis
- C. Hypercalciuria
- D. Cysteine

13. _____ cell wall develops between plasma membrane and primary cell wall.

- A. Middle lamella
- B. Tertiary
- C. Secondary
- D. Elongated

14. Movement of substances from a region of high concentration to low concentration is called _____

- A. Effusion
- B. Osmosis
- C. Rapid flow
- D. Filtration

15. In human beings, lung volume can be measured by special instrument called _____

- A. Vital pipe
- B. Spectrometry
- C. Spectroscopy
- D. Spirometry

16. The process of joining two monomers by removing water molecule is called _____

- A. Condensation
- B. Hydrolysis
- C. Hydration
- D. Decomposition

17. The body sometime eliminates old cells or unwanted cells at embryonic stage according to their genetic information, this process is called _____

- A. Apoptosis
- B. Autolysis
- C. Krabbes
- D. Gauche

18. In humans _____ % uric acid is

19. The inflammation of the air passage ways is known as _____

- A. Pneumonia
- B. Asthma
- C. Bronchitis
- D. Lung cancer

20. The total surface area of the respiratory surface in human is about _____ times to the size of the body.

- A. 10
- B. 20
- C. 15
- D. 25

21. _____ enzyme act on amylose to convert it into maltose.

- A. Maltase
- B. Lipase
- C. Amylase
- D. Lactase

22. The word Botany is taken from Greek language _____

- A. Butane
- B. Biotine
- C. Botaline
- D. Butanase

23. Plants exchange gases for the process of photosynthesis and _____

- A. Respiration
- B. Breathing

24. The overall muscle organ is surrounded by membrane called _____

- A. Peritonium
- B. Endomysium
- C. Fascicle
- D. Epimysium

25. The solution of biological problem starts with _____

- A. Experiment
- B. Hypothesis
- C. Observation
- D. Reasoning

26. Human nervous system consists of _____ parts.

- A. 4
- B. 2
- C. 5
- D. 3

27. In human body, there are _____ pairs of cranial nerves.

- A. 15
- B. 20
- C. 12
- D. 31

28. _____ is a junction that controls communication between a neuron and another cell.

- A. Sclerotomy
- B. Synapse
- C. Signal

29. The branch of science which deals with the study of human bones is called _____

- A. Histology
- B. Morphology
- C. Cytology
- D. Osteology

30. The study of structure of tissue of plant and animals is known as _____

- A. Taxonomy
- B. Palaeontology
- C. Histology
- D. Physiology

31. The term homeostasis was introduced by _____

- A. Walter
- B. Fleming
- C. Mendel
- D. Joseph

32. Each enzyme has special charged place to fit the substrate called _____

- A. Coenzyme
- B. Active site
- C. Apo-enzyme
- D. Holoenzyme

33. _____ refers to the difficulty in focusing distance objects.

- A. Hyperopia
- B. Strabismus
- C. Myopia
- D. Mydriasis

34. Short chain amino acid hormones are called _____ hormone.

- A. Polypeptide
- B. Peptide
- C. Biogenetic
- D. Tripeptide

35. Water is a _____ molecule.

- A. Non-polar
- B. Polar
- C. Covalent
- D. Ionic

36. The water which contains very low amount of salt is called _____

- A. Standing water
- B. Running water
- C. Sea water
- D. Fresh water

37. Which one of the following is NOT included in five kingdom system?

- A. Bacteria
- B. Amoeba
- C. Fungi
- D. Virus

38. Micro-filament is solid strands of about _____ nm in diameter.

- A. 9
- B. 2
- C. 7
- D. 10

39. There are _____ types of enzyme inhibition.

40. Golgi body consist of central flattened plate like compartments called _____

- A. Polysome
- B. Cisternae
- C. Canaliculi
- D. Microtubules

PHYSICS

41. Least count is _____ the precision of measurement equipment.

- A. Independent of
- B. Directly proportional
- C. Inversely proportional
- D. None of these

42. At what angle between two vectors scalar product is maximum?

- A. 30°
- B. 60°
- C. 0°
- D. 45°

43. If we double the mass and half force, then acceleration produced will be

- A. Double
- B. Four times
- C. Remains same
- D. One fourth

44. A collision in which both kinetic energy and momentum is conserved is called

45. If we double the magnitude of charges and double the distance between charges then the magnitude of force will become
- Sixteen times
 - Four times
 - Eight times
 - Remains same
46. At what angle between two vectors, their scalar product is equal to their vector product?
- 30°
 - 60°
 - 0°
 - 45°
47. Impulse is equal to
- momentum
 - Change in momentum
 - Product of force and time
 - Both B and C
48. Ammeter is connected in _____ in circuit.
- Series
 - Parallel
 - Series and parallel
 - None of these
49. The shape of the falling drop is due to
- Viscosity
 - Surface tension
 - Buoyancy
 - Uplift force
50. If we follow the electric field lines due to positive point charge then electric field strength will
- Increase
 - Decrease
 - Remains same
 - None of these
51. If we double the area of plates and half the distance between the plates then the capacitance of a capacitor will become
- Double
 - Half
 - Four times
 - One fourth
52. According to which of the following law, volume of gas is directly proportional to temperature at constant pressure.
- Boyle's Law
 - Charles Law
 - Avogadro's Law
 - Graham's Law of diffusion
53. Torque acting on current carrying coil placed in uniform magnetic is minimum when plane of coil is placed _____.
- Parallel to the magnetic field
 - Perpendicular to the magnetic field
 - 45° to the magnetic field
 - 60° to the magnetic field
54. Torque acting on a current carrying coil is the working principle of
- Moving coil galvanometer
 - Wheatstone bridge
 - Meter bridge
 - Potentiometer
55. Which one the following is dimensionless quantity?
- Pressure
 - Angle
 - Velocity
 - Force
56. Electric flux through a close surface enclosing an electric dipole is
- Positive
 - Negative
 - Zero
 - Either positive or negative
57. Which material has a negative temperature coefficient of resistance?
- Iron
 - Mercury
 - Silicon
 - Constantan
58. If a wire of resistance R is stretched three time, then new resistance of wire will become
- R
 - 3R
 - 9R
 - R/3
59. The value of escape velocity is independent of
- Mass of object
 - Mass of planet
 - Radius of planet
 - Both B and C
60. If a ball is thrown vertically upward with velocity 10m/s then maximum height attained will be
- 5 m
 - 10 m
 - 15 m
 - 20 m
61. If we double the potential difference between the plates of charged capacitor then energy stored will become _____ of initial value.
- Double
 - Half
 - Four times
 - One fourth
62. Voltmeter is connected in _____ in circuit.
- Series
 - Parallel
 - Series and parallel
 - None of these
63. According to Pascal's Law, an increase in pressure at any point in an enclosed fluid is transmitted
- In opposite direction
 - In same direction
 - In perpendicular direction
 - In all direction

64. The relation between time period and orbital radius for satellite is given by

- A. Kepler's First Law
- B. Kepler's Second Law
- C. Kepler's Third Law
- D. Kepler's Fourth Law

65. In transformer, which of the following quantity remain same during mutual induction?

- A. Voltage
- B. Current
- C. Frequency
- D. Both A and B

66. The negative of Potential gradient is equal to

- A. Electric field intensity
- B. Potential difference
- C. Work done
- D. Charge

67. Which of the following quantity is base?

- A. Force
- B. Mass
- C. Velocity
- D. Acceleration

68. The slope of momentum time graph is equal to

- A. Power
- B. Impulse
- C. Acceleration
- D. force

69. Which of the following quantity is supplementary?

- A. Power
- B. Weight
- C. Radian
- D. Mass

70. The dot product of force and velocity is

- A. Mass
- B. Energy
- C. Power
- D. Work

CHEMISTRY

71. An orbital can have maximum two electrons with opposite spin according to

- A. Hund's rule
- B. Pauli exclusion principle
- C. Aufbau principle
- D. Heisenberg principle

72. Alpha rays have high ionization power due to

- A. Small size
- B. Large size
- C. Have no charge
- D. Have negative charge

73. The number of covalent bond present in 12g of NH_3 are:

- A. 1.4×10^{24}
- B. 1.2×10^{24}
- C. 1.52×10^{23}
- D. 1.53×10^{24}

74. 46g of ethyl alcohol and 180g of glucose having same

- A. Number of atoms
- B. Same number of electrons
- C. Same number of protons
- D. Same number of molecules

75. Paschen, Bracket and Pfund series lies in the

- A. Visible region
- B. Microwave region
- C. Infrared region
- D. Ultraviolet region

76. Number of moles in 35g of sodium metal are

- A. 1.52g
- B. 1.90g
- C. 1.32g
- D. 2.20g

77. The quantum number which explain shape of orbital is

- A. Spin quantum number
- B. Magnetic quantum number
- C. Principle quantum number
- D. Azimuthal quantum number

78. Electron is how much less than neutron

- A. 1836 times
- B. 1842 times
- C. 1839 times
- D. 1830 times

79. Example of colloids containing gas in liquid is

- A. Milk
- B. Jelly
- C. Fog
- D. Rubby glass

80. 27g of Al will react with how much mass of O_2 to produce Al_2O_3

- A. 8g of O_2
- B. 12g of O_2
- C. 36g of O_2
- D. 24g of O_2

81. Stoicheion is a Greek word which means

- A. Element
- B. Molecules
- C. Atom
- D. Ion

82. Maximum number of molecules are present in at STP

- A. 22.414dm^3 of O_2
- B. 1 moles of O_2
- C. 44g of CO_2
- D. 64g of O_2

83. James Chadwick in 1932 AD discovered

- A. Proton
- B. Neutron
- C. Electron
- D. α - particle



84. Theory which successfully explain paramagnetic behavior of O_2 molecules
- VSEPR theory
 - MOT
 - VBT
 - Quantum theory
85. A substance which consumed earlier in a chemical reaction is called
- Excess reactant
 - Actual yield
 - Limiting reactant
 - Unreacting reactant
86. The value of Planck's constant is
- $6.62 \times 10^{-34} \text{ J.s}$
 - $6.62 \times 10^{-26} \text{ J.s}$
 - $6.62 \times 10^{-24} \text{ J.s}$
 - $6.62 \times 10^{-27} \text{ J.s}$
87. 12g of carbon and 24g of magnesium having same
- Number of molecules
 - Atomic number
 - Mass number
 - Number of atoms
88. Measure of internal resistance of a liquid is called
- Surface tension
 - Evaporation
 - Boiling point
 - viscosity

89. Experimentally determined yield is also called:

- Theoretical yield
- Percent yield
- Actual yield
- Manufacturing yield

90. The mass of 15dm^3 of O_2 is equal to

- 20g
- 21.41g
- 32g
- 76g

91. Which statement is correct for 1 mole of CO_2

- 22.414dm^3 volume of STP
- 44g of CO_2
- 6.022×10^{23} molecules of STP
- 88g of CO_2

92. $n + 1$ rule is used for

- Shape of orbitals
- Filling of electrons in orbitals
- Energy of orbitals
- Shape of orbit

93. The study of the absorption and emission of light by matter is called:

- Endoscopy
- Spectroscopy
- Excitation
- De-excitation

94. Which statement is incorrect about 24g of carbon dioxide (CO_2)

- 1.2044×10^{24} molecules of CO_2
- 2 moles of CO_2
- 32g of oxygen
- 22.4dm^3 of CO_2 at STP

95. Calculate number of moles of 90g of water

- 5 moles
- 2.5 moles
- 10 moles
- 18 moles

96. Which one of the following has formula mass

- $CaH_{12}O_6$
- KCl
- CO_2
- NH_3

97. Third line of Balmer's series is

- $n_1=2, n_2=5$
- $n_1=2, n_2=4$
- $n_1=2, n_2=5$
- $n_1=3, n_2=5$

98. Degenerate orbitals are orbitals of

- Different energy
- Same energy
- Increase in energy
- Decrease in energy

99. Bond order of O_2 molecules is

- 2
- 3
- 4
- Zero order

100. Calculate energy of electrons when $n=2$ in K /mole

- -328.25 K J/mole
- -479.4 K J/mole
- -632.4 K J/mole
- -812.2 K J/mole

